

169.99
169.94

155.63

65.05

48.50
48.25

32.62
32.52

Int= average of normalized integrals values

MW =molecular weight

P =Purity (as percent value)

m=mass

n= number of carbon giving rise to a given NMR signal

$n_2=1$

$Int_2= 1.1004$

$MW_2= 301.97$

$m_2= 72.2$ mg

$\mu\text{mol} = 239$

$n_{EC}=1$

$Int_{EC}= 1.0000$

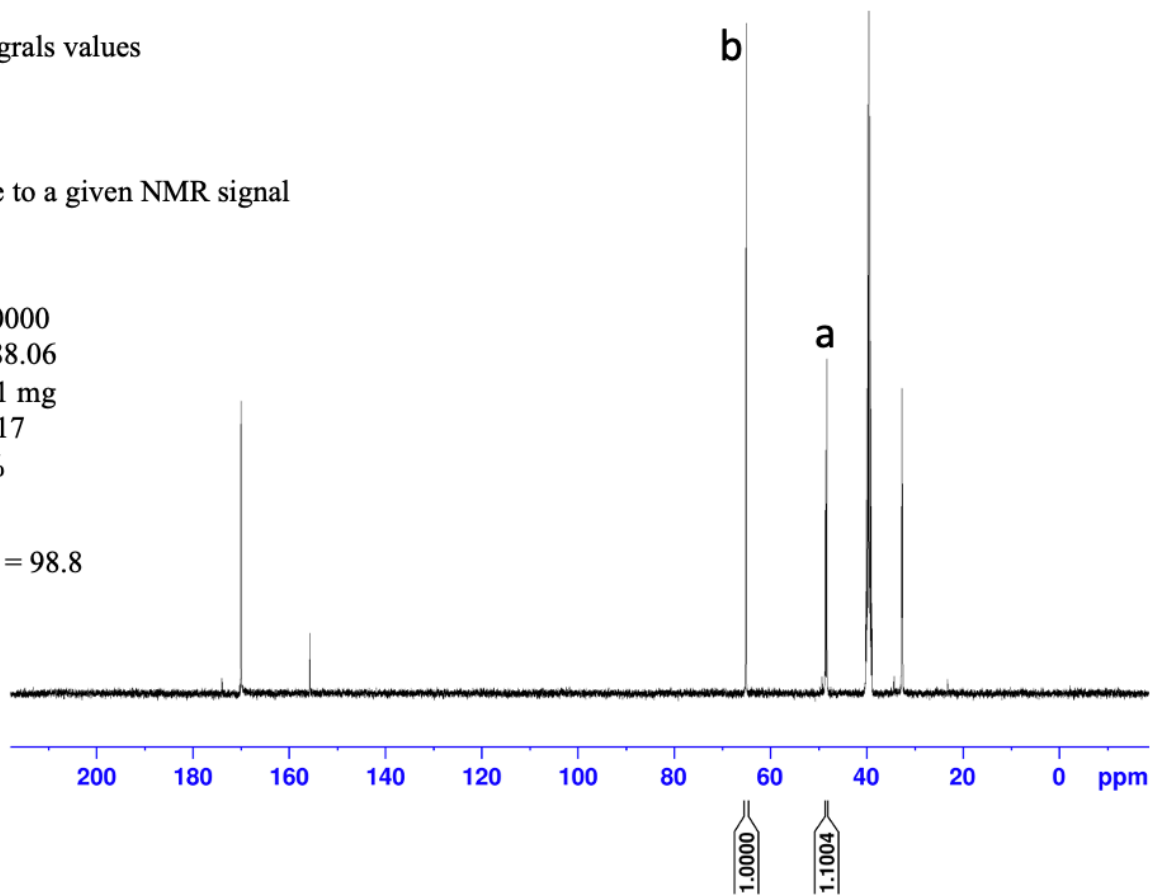
$MW_{EC}= 88.06$

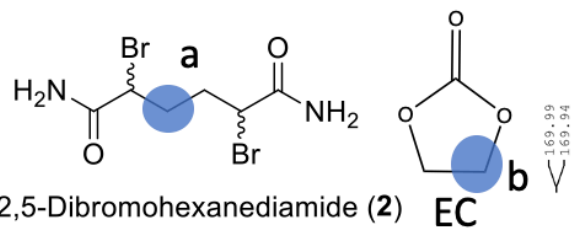
$m_{EC}= 19.1$ mg

$\mu\text{mol} = 217$

$P_{EC} >99\%$

$$P(\%) = \left(\frac{n_{EC} \cdot Int_2 \cdot MW_2 \cdot m_{EC}}{n_2 \cdot Int_{EC} \cdot MW_{EC} \cdot m_2} \right) \cdot P_{EC} = 98.8$$





169.99
 169.94
 155.63

65.06
 48.51
 48.25
 32.62
 32.52

Int= average of normalized integrals values
 MW =molecular weight
 P =Purity (as percent value)
 m=mass
 n= number of carbon giving rise to a given NMR signal

$n_2=1$	$n_{EC}=1$
$Int_2= 1.1048$	$Int_{EC}= 1.00$
$MW_2= 301.97$	$MW_{EC}= 88.06$
$m_2= 73.5 \text{ mg}$	$m_{EC}= 19.1 \text{ mg}$
$\mu\text{mol} = 240$	$\mu\text{mol} = 217$
	$P_{EC} >99\%$

$$P(\%) = \left(\frac{n_{EC} \cdot Int_2 \cdot MW_2 \cdot m_{EC}}{n_2 \cdot Int_{EC} \cdot MW_{EC} \cdot m_2} \right) \cdot P_{EC} = 97.5$$

